

# STIC Search Report Biotech-Chem Library

# STIC Database Tracking Number: 104335

TO: Deborah Lambkin

Location:

Art Unit: 1626

September 23, 2003

Case Serial Number: 09/937632

From: P. Sheppard Location: CM1-1E03 Phone: (703) 308-4499

sheppard@uspto.gov

# Search Notes

## SEARCH REQUEST FORM

Scientific and Technical Information Center

An Lint 1626 Phone Number 30 8 45 Phone Number 30 8 Phone Number 3	Requester's Full Name:	Lambler	1:xaminer = : 7/300 Date: 2779776	
If more than one search is submitted, please prioritize searches in order of need.  Please provide a decaded statement of the search topic, and describe as specifically as possible the subject matter to be searched include the elected species or suncures, keywords, synonyms, actionyms, and registry numbers, and combine with the concept or unitive of the nevertino Define any terms that may have a special meaning. Once examples or relevant citations, authors, etc., if known Please attach a copy of the cover sheet, perinent claims, and abstract.  Title of Invention:  Fire Normal Please provide full names:  Draylos Brukett  Earliest Priority-Filing Date:  "For Sequence Searches Only" Please include all perinent information (parent, child, divisional, or issued patent numbers) along with the appropriate serial number.  H, CH3, NH  R= methyl or Calcus  H, CH3, NH  R= Methyl or Calcus  NH  Please attach or serial numbers.  The CH3 NH  R= Methyl or Calcus  NH  Please include all perinent information (parent, child, divisional, or issued patent numbers) along with the appropriate retrial number.  The CH3, NH  R= Methyl or Calcus  NH  Please include all perinent information (parent, child, divisional, or issued patent numbers) along with the appropriate retrial number.  The CH3, NH  R= Methyl or Calcus  NH  Plans Define Search  NA Sequence (see Search Search Vendors and cost where applicable search Sear	An Unit. 1626 Phor	ne Number 30 8-452	2 Serial Number: <u>09/977, 632</u>	
Please provide a deciated statement of the search topic, and describe as specifically as possible the subsect matter to be searched include the elected secrets or structures, keywords, synonyms, acronyms, and registry numbers, and combine with the concept or uniting of the incomon Define any terms that may have a special meaning. Give examples or relevant citations, authors, etc., if known Please attach a copy of the cover sheet, perinnent claims, and abstract.  Title of Invention:  Fine Maps Habris Cared:  Inventors iplease provide full names):  Draylos Sumbett  Earliest Priority Filing Date:  For Sequence Searched Only? Please include all pertinent information (parent, child, divisional, or issued patent numbers) along with the apparaments resist number.  A CA3 NA — R  WHY CA3 NA — R  Paula, Search appropriate and appropriate and appropriate activities of the control of the country of the co	Mail Box and Bidg/Room Local	lion: <u>(M/3603</u> Ro	sults Format Preferred (circle): PAPER DISK E-MAIL	
Include the elected secrets or structures, keywords, syronyms, and registry numbers, and combine with the concept or within for the more please attach a copy of the cover sheet, perinent claims, and abstract.  Title of Invention:  IN Upo Setts Ordel.  Inventions (please provide full names):  Douglas Brukett  Earliest Priority-Filing Date:  "For Sevence Searches Only" Please include all pertinent information (parent, child, divisional, or issued patent numbers) along with the appropriate verial number.  R = method or Color NAT	If more than one search is su	bmitted, please priorit	tize searches in order of need.	
withing of the invention. Define any terms that may have a special meaning. Give examples or relevant citations, authors, etc., if known Please afficials a copy of the cover sheet, permiter claims, and sharted:  Title of Invention:  IN Upo State Cradical  Inventions (please provide full names):  Descriptors  Buildett  Earliest Priority-Filing Date:  "For Sequence Searches Only" Please include all perinent information (parent, child, divisional, or issued parent numbers) along with the apparamental verial number.  H, CH3, NH  R = Methyl  A CH3  WH2  CH3  WH2  CH3  WH3  Start Descriptors  STAFF USE ONLY  Type of Search  A Sequence (n)  STAFF USE ONLY  Type of Search  Street Please  A Sequence (n)  Street Please  Street Please  Street Please Street  Bibliographic  Server Boase Street  Please Family  WWW Interest  Server System  Please Family  WWW Interest  Server System  Please Family  WWW Interest  Server Systems  Server Systems  Please Family  WWW Interest  Server Systems  Serve	Please provide a detailed statement of	the search topic, and describ	e as specifically as possible the subject matter to be searched	
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Inventors iplease provide full names): Dragles Burkett  Earliest Priority Filing Date:  **For Sequence Searches Only** Please include all pertinent information (parent, child, divisional, or issued parent numbers) along with the apparamental verial number.  **H, CH3, NH  **NH CH3 or Galax  **NH CH	known. Please attach a copy of the cov	ver sheet, pertinent claims, ar	nd abstract.	
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FILE COVERS 1907 - 23 Sep 2003 VOL 139 ISS 13 FILE LAST UPDATED: 22 Sep 2003 (20030922/ED)

This file contains CAS Registry Numbers for easy and accurate substance identification.

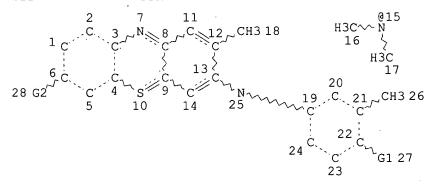
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NODE ATTRIBUTES: DEFAULT MLEVEL IS ATOM DEFAULT ECLEVEL IS LIMITED

GRAPH ATTRIBUTES: RING(S) ARE ISOLATED OR EMBEDDED NUMBER OF NODES IS 14

STEREO ATTRIBUTES: NONE

L7 31130 SEA FILE=REGISTRY SSS FUL L5 L12 STR



Lambkin 10 937632 VAR G1=NH2/CH3 VAR G2=NH2/15 NODE ATTRIBUTES: DEFAULT MLEVEL IS ATOM DEFAULT ECLEVEL IS LIMITED GRAPH ATTRIBUTES: RSPEC I NUMBER OF NODES IS 28 · STEREO ATTRIBUTES: NONE 2 SEA FILE=REGISTRY SUB=L7 SSS FUL L12 L13100.0% PROCESSED 3 ITERATIONS ·2 ANSWERS SEARCH TIME: 00.00.01 => => s 1131 L13 L14=> => => d ibib abs hitrn 114 1-YOU HAVE REQUESTED DATA FROM 1 ANSWERS - CONTINUE? Y/(N):n => d ibib abs hitrn 114 1 L14 ANSWER 1 OF 1 HCAPLUS COPYRIGHT 2003 ACS on STN ACCESSION NUMBER: 2001:564838 HCAPLUS 135:134287 DOCUMENT NUMBER: TITLE: In vivo stain compounds and methods of use to identify dysplastic tissue Burkett, Douglas D. INVENTOR(S): PATENT ASSIGNEE(S): Zila, Inc., USA PCT Int. Appl., 51 pp. SOURCE: CODEN: PIXXD2 DOCUMENT TYPE: Patent LANGUAGE: English FAMILY ACC. NUM. COUNT: 1 PATENT INFORMATION:

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AB Compds. having the structural formula I wherein X is hydrogen, Me, or Y; Y is -NH-R or hydrogen; and R is Me or formula II are useful as in vivo stains for the detection of dysplastic tissue.

IT 352005-62-0P 352005-65-3P

RL: BUU (Biological use, unclassified); SPN (Synthetic preparation); BIOL (Biological study); PREP (Preparation); USES (Uses)

(In vivo stain compds. and methods of use to identify dysplastic tissue)

REFERENCE COUNT:

THERE ARE 3 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

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=> fil caold

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FILE COVERS 1907-1966 FILE LAST UPDATED: 01 May 1997 (19970501/UP)

This file contains CAS Registry Numbers for easy and accurate substance identification. Title keywords, authors, patent assignees, and patent information, e.g., patent numbers, are now searchable from 1907-1966. TIFF images of CA abstracts printed between 1907-1966 are available in the PAGE display formats.

This file supports REG1stRY for direct browsing and searching of all substance data from the REGISTRY file. Enter HELP FIRST for more information.

=> => => s 113 L15

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Property values tagged with IC are from the ZIC/VINITI data file

provided by InfoChem.

STRUCTURE FILE UPDATES: 22 SEP 2003 HIGHEST RN 591204-55-6 DICTIONARY FILE UPDATES: 22 SEP 2003 HIGHEST RN 591204-55-6

TSCA INFORMATION NOW CURRENT THROUGH JULY 14, 2003

Please note that search-term pricing does apply when conducting SmartSELECT searches.

Crossover limits have been increased. See HELP CROSSOVER for details.

Experimental and calculated property data are now available. See HELP PROPERTIES for more information. See STNote 27, Searching Properties in the CAS Registry File, for complete details: http://www.cas.org/ONLINE/STN/STNOTES/stnotes27.pdf

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L13 ANSWER 1 OF 2 REGISTRY COPYRIGHT 2003 ACS on STN

RN 352005-65-3 REGISTRY

CN Phenothiazin-5-ium, 7-amino-3-[(4-amino-3-methylphenyl)amino]-2-methyl-(9CI) (CA INDEX NAME)

FS 3D CONCORD

MF C20 H19 N4 S

SR CA

LC STN Files: CA, CAPLUS

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1 REFERENCES IN FILE CA (1907 TO DATE)

1 REFERENCES IN FILE CAPLUS (1907 TO DATE)

REFERENCE 1: 135:134287

L13 ANSWER 2 OF 2 REGISTRY COPYRIGHT 2003 ACS on STN

RN 352005-62-0 REGISTRY

CN Phenothiazin-5-ium, 3-[(4-amino-3-methylphenyl)amino]-7-(dimethylamino)-2-methyl- (9CI) (CA INDEX NAME)

FS 3D CONCORD

MF C22 H23 N4 S

SR CA

LC STN Files: CA, CAPLUS

$$Me_2N$$
 $NH_2$ 
 $NH_2$ 
 $NH_2$ 

1 REFERENCES IN FILE CA (1907 TO DATE)
1 REFERENCES IN FILE CAPLUS (1907 TO DATE)

REFERENCE 1: 135:134287

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FILE COVERS 1907 - 23 Sep 2003 VOL 139 ISS 13 FILE LAST UPDATED: 22 Sep 2003 (20030922/ED)

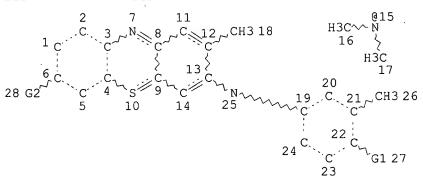
This file contains CAS Registry Numbers for easy and accurate substance identification.

NODE ATTRIBUTES: DEFAULT MLEVEL IS ATOM DEFAULT ECLEVEL IS LIMITED

GRAPH ATTRIBUTES: RING(S) ARE ISOLATED OR EMBEDDED. NUMBER OF NODES IS 14

STEREO ATTRIBUTES: NONE

L7 31130 SEA FILE=REGISTRY SSS FUL L5 L12 STR



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VAR G1=NH2/CH3
VAR G2=NH2/15
NODE ATTRIBUTES:
DEFAULT MLEVEL IS ATOM
DEFAULT ECLEVEL IS LIMITED
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GRAPH ATTRIBUTES:

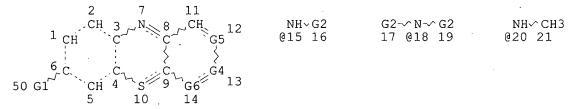
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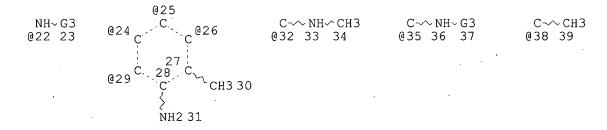
NUMBER OF NODES IS 28

STEREO ATTRIBUTES: NONE

2 SEA FILE=REGISTRY SUB=L7 SSS FUL L12 L13

L20 STR





C~~NH~CH3  $C \sim NH$ C-√ CH3  $C \sim N \sim G3$ @42 43 044 45 46 @47 48 49 040 41

VAR G1=NH2/15/18 VAR G2=CH3/20/22 VAR G3=24/25/26/29 VAR G4=CH/32/35 VAR G5=CH/38/40 VAR G6=CH/42/44/47 NODE ATTRIBUTES: DEFAULT MLEVEL IS ATOM

DEFAULT ECLEVEL IS LIMITED

GRAPH ATTRIBUTES:

RING(S) ARE ISOLATED OR EMBEDDED

NUMBER OF NODES IS 50

STEREO ATTRIBUTES: NONE

L21 60 SEA FILE=REGISTRY SUB=L7 SSS FUL L20 L22 SCR 2127

L23 12 SEA FILE=REGISTRY SUB=L21 SSS FUL L20 NOT L22 10 SEA FILE=REGISTRY ABB=ON PLU=ON L23 NOT L13 L24

20 SEA FILE=HCAPLUS ABB=ON PLU=ON L24 L25

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L25 ANSWER 1 OF 20 HCAPLUS COPYRIGHT 2003 ACS on STN

ACCESSION NUMBER:

2002:226744 HCAPLUS

DOCUMENT NUMBER:

136:226765

TITLE:

Phenothiazinium derivative having an antiparasitic and

biological activity

INVENTOR(S):

Galey, Laurent

PATENT ASSIGNEE(S):

Fr.

SOURCE:

Fr. Demande, 13 pp.

CODEN: FRXXBL

DOCUMENT TYPE:

Patent

LANGUAGE:

French

FAMILY ACC. NUM. COUNT:

1

PATENT INFORMATION:

APPLICATION NO. PATENT NO. KIND DATE \_\_\_\_\_ \_\_\_\_ \_\_\_\_\_ FR 2810549 20011228 FR 2000-8249 A1 PRIORITY APPLN. INFO.: FR 2000-8249

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DATE

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AΒ Pharmaceutical compn. with antiparasitic activity against Plasmodium, Babesia, Toxoplasma, Trypanosoma, Onchocerca, Filaria, Leishmania, Nematodes, Plathelminthines, and Nemathelminthes are disclosed comprising the structure of I (no data).

29260-45-5 IT

> RL: PAC (Pharmacological activity); THU (Therapeutic use); BIOL (Biological study); USES (Uses) (phenothiazinium deriv. having antiparasitic and biol. activity)

L25 ANSWER 2 OF 20 HCAPLUS COPYRIGHT 2003 ACS on STN

ACCESSION NUMBER:

2001:564838 HCAPLUS

DOCUMENT NUMBER:

135:134287

TITLE:

In vivo stain compounds and methods of use to identify

dysplastic tissue Burkett, Douglas D.

PATENT ASSIGNEE(S): SOURCE:

Zila, Inc., USA PCT Int. Appl., 51 pp.

CODEN: PIXXD2

DOCUMENT TYPE:

INVENTOR(S):

Patent

LANGUAGE:

English

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

KIND DATE APPLICATION NO. DATE PATENT NO. 20000131 WO 2001054696 A1 20010802 WO 2000-US2602

. W: AU, BR, CA, CN, CZ, HU, IL, IN, JP, KR, MX, NO, PL, SG, SK, TR,

US, ZA, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM

RW: AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL,

PT, SE

EP 1165087 A1 20020102 EP 2000-915730 20000131

R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT,

IE, FI

BR 2000009427 A 20020716 BR 2000-9427 20000131 JP 2003520816 T2 20030708 JP 2001-554680 20000131 NO 2001004720 A 20011127 NO 2001-4720 20010928

NO 2001004720 A 20011127 NO 2001-4720 20010928 PRIORITY APPLN. INFO.: WO 2000-US2602 W 20000131

OTHER SOURCE(S): MARPAT 135:134287

GΙ

AB Compds. having the structural formula I wherein X is hydrogen, Me, or Y; Y is -NH-R or hydrogen; and R is Me or formula II are useful as in vivo stains for the detection of dysplastic tissue.

IT 47078-64-8P 352005-59-5DP, derivs. 352005-60-8P
352005-61-9P 352005-63-1P

RL: BUU (Biological use, unclassified); SPN (Synthetic preparation); BIOL (Biological study); PREP (Preparation); USES (Uses)

(In vivo stain compds. and methods of use to identify dysplastic

tissue)

REFERENCE COUNT: 3 THERE ARE 3 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L25 ANSWER 3 OF 20 HCAPLUS COPYRIGHT 2003 ACS on STN

ACCESSION NUMBER: 1999:640092 HCAPLUS

DOCUMENT NUMBER: 131:334255

TITLE: Investigation on the iodination reaction of methylene

blue by liquid chromatography-mass spectrometry with

ionspray ionization

AUTHOR(S): Raffaelli, Andrea; Pucci, Sergio; Desideri, Ielizza;

Bellina, Calogero R.; Bianchi, Romano; Salvadori,

Piero

CORPORATE SOURCE: Centro di Studio del CNR per le Macromolecole

Stereordinate ed Otticamente Attive, Dipartimento di

Chimica e Chimica Industriale, Universita di Pisa,

Pisa, 56126, Italy

SOURCE: Journal of Chromatography, A (1999), 854(1 + 2), 57-67

CODEN: JCRAEY; ISSN: 0021-9673

PUBLISHER: Elsevier Science B.V.

DOCUMENT TYPE: Journal LANGUAGE: English

AB Radioactive iodine (131I and 123I) labeled methylene blue is used for the early diagnosis of melanoma metastases. We studied the iodination reaction of methylene blue (using "cold" iodine) in order to characterize the iodination product(s) as far as no. and position of iodine atoms introduced on the arom. ring(s) is concerned. The reaction was carried out under the same exptl. conditions used for the radioactive one, that is in a large excess of methylene blue. The ion-spray HPLC-MS anal. of the reaction mixt. showed that the iodinated methylene blue was present only in a very small amt. and the main iodinated product was a demethylated one, coming out from the iodination of an impurity azure B. We also

studied the iodination reaction of azure B in order to better explain the reaction pathway. Com. azure B contains impurities of methylene blue and all the possible demethylated derivs. HPLC-MS anal. of the reaction mixt. allowed a complete characterization of the iodinated and bis-iodinated

29260-45-5 30719-07-4 39093-22-6 ΤТ

> RL: RCT (Reactant); THU (Therapeutic use); BIOL (Biological study); RACT (Reactant or reagent); USES (Uses)

(investigation on the iodination reaction of methylene blue by liq.

chromatog.-mass spectrometry with ion-spray ionization)

29260-45-5DP, iodinated 30719-07-4DP, iodinated 39093-22-6DP, iodinated IT

RL: SPN (Synthetic preparation); THU (Therapeutic use); BIOL (Biological study); PREP (Preparation); USES (Uses)

(investigation on the iodination reaction of methylene blue by liq.

chromatog.-mass spectrometry with ion-spray ionization)
E COUNT: 12 THERE ARE 12 CITED REFERENCES AVAILABLE FOR THIS REFERENCE COUNT: RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L25 ANSWER 4 OF 20 HCAPLUS COPYRIGHT 2003 ACS on STN

ACCESSION NUMBER:

1996:702983 HCAPLUS

DOCUMENT NUMBER:

125:341110

TITLE:

Electrochemical study of the reaction between progressively alkylated thiazine leuco dyes and

Fe(III) on a glassy carbon electrode

AUTHOR(S):

Ahmed, S.; Saha, S. K.

CORPORATE SOURCE:

Dep. Chem., Univ. North Bengal, Darjeeling, 734430,

India

SOURCE:

Canadian Journal of Chemistry (1996), 74(10),

1896-1902

CODEN: CJCHAG; ISSN: 0008-4042

PUBLISHER:

National Research Council of Canada

DOCUMENT TYPE: LANGUAGE:

Journal English

An electrochem. study on five progressively alkylated thiazine dyes in the presence of Fe(III) ions is reported. The theory of the catalytic regeneration mechanism involving an electrode reaction followed by a coupled chem. reaction is applied to derive kinetic parameters of homogeneous reaction. The 2nd-order rate const. for the reaction of thiazine leuco dyes with Fe(III) ions increases from 0.25 .times. 104 to 1.6 .times. 104 dm3 mol-1 s-1 upon monomethylation and to vary from 0.7 .times. 104 for the di-Me deriv. to 1.4 .times. 104 dm3 mol-1 s-1 for the tetra-Me one. The electron-donating nature as well as the hydrophobic characteristics of the Me group influence the kinetics of the homogeneous reaction.

ΙT 29260-45-5P 30719-07-4P

RL: PEP (Physical, engineering or chemical process); PRP (Properties); PUR (Purification or recovery); RCT (Reactant); PREP (Preparation); PROC (Process); RACT (Reactant or reagent)

(cyclic voltammetry in sulfuric acid: electrochem. study of reaction between progressively alkylated thiazine leuco dyes and Fe(III) on glassy carbon electrode)

L25 ANSWER 5 OF 20 HCAPLUS COPYRIGHT 2003 ACS on STN

ACCESSION NUMBER: 1996:631422 HCAPLUS

DOCUMENT NUMBER: 125:329347

TITLE: Receptor inactivation by dye-neuropeptide conjugates:

1. The synthesis of Cys-containing dye-neuropeptide

AUTHOR(S): Feigenbaum, Jeffery J.; Choubal, Milind D.; Payza,

Kemal; Kanofsky, Jeffrey R.; Crumrine, David S.

Department of Research and Development, American Inst. CORPORATE SOURCE:

of Biotechnology, Elk Grove, IL, 60007, USA

SOURCE: Peptides (Tarrytown, New York) (1996), 17(6), 991-994

CODEN: PPTDD5; ISSN: 0196-9781

PUBLISHER: Elsevier
DOCUMENT TYPE: Journal
LANGUAGE: English

AB To attenuate specifically identified receptors through photolysis, a four-step synthesis of a useful tethered deriv. of Azure-B (Az) was developed. After characterization, this, deriv. was covalently attached to CFMREamide, CFMRF, and CLRFamide (i.e., three different neuropeptide analogs of the putative neurotransmitter FMRFamide). This resulted in the formation of three dye-neuropeptide conjugates: Az-CFMRFamide, Az-CFMRF, and Az-CLRFamide.

IT 29260-45-5P

RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(synthesis of Cys-contq. dye-neuropeptide conjugates)

L25 ANSWER 6 OF 20 HCAPLUS COPYRIGHT 2003 ACS on STN

ACCESSION NUMBER: 1996:229160 HCAPLUS

DOCUMENT NUMBER: 124:328070

TITLE: Electrochemichromic solutions, processes for preparing

and using the same, and devices manufactured with the

same

INVENTOR(S): Varaprasad, Desaraju V.; Looman, Steven D.; Zhao,

Mingtang; Habibi, Hamid R.; Lynam, Niall R.

PATENT ASSIGNEE(S): Donnelly Corp., USA

SOURCE: U.S., 32 pp., Cont.-in-part of U.S. 5, 239, 405.

CODEN: USXXAM

DOCUMENT TYPE: Patent LANGUAGE: English

FAMILY ACC. NUM. COUNT: 2

PATENT INFORMATION:

PAT	ENT NO.	KIND	DATE		APPLICATION NO.	DATE
US EP EP	5500760 5239405 531143 531143 R: DE, FR 07216349	A A A2 A3 , GB, IE,	19960319 19930824 19930310 19931020 IT 19950815		US 1992-935784 US 1991-756342 EP 1992-308022 JP 1992-238612	19920827 19910906 19920904
US US US	5424865 5611966 5985184 6143209	A A A A	19950613 19970318 19991116 20001107		US 1993-61742 US 1995-458080 US 1997-956198 US 1999-325712	19930117 19950601 19971022 19990604
PRIORITY	APPLN. INF	0.:		US US US US US US	1991-756342 A2 1992-935784 A 1992-308022 W 1993-61742 A3 1995-458080 A3 1997-819652 B3 1997-956198 A3	19920827 19920904 3 19930117 3 19950601 19970317

OTHER SOURCE(S): MARPAT 124:328070

AB Electrochemichromic solns. capable of color change when a potential is applied comprise at least one anodic compd., said anodic compd. having been previously contacted with a redox agent such that said anodic compd. exists in a different valence state than prior to having been contacted with said redox agent, at least one cathodic compd., and a solvent wherein the redox potential of the anodic compd. in the different valence state is greater than the redox potential of the cathodic compd. while in contact with the solvent. Electrochemichromic devices (e.g., mirrors, glazings, partitions, filters, displays, and lenses) employing the solns. in a cell are also described.

### IT 29260-45-5 30719-07-4

RL: DEV (Device component use); TEM (Technical or engineered material use); USES (Uses)

(electrochemichromic solns. using prereduced anodic compds. and devices using them)

L25 ANSWER 7 OF 20 HCAPLUS COPYRIGHT 2003 ACS on STN

ACCESSION NUMBER: 1994:229898 HCAPLUS

DOCUMENT NUMBER: 120:229898

TITLE: Electrochemichromic solutions, processes for preparing

and using the same, and devices manufactured with the

same

INVENTOR(S): Varaprasad, Desaraju V.; Habibi, Hamid R.; Looman,

Steven D.; Lynam, Niall R.; Zhao, Mingtang

PATENT ASSIGNEE(S): Donnelly Corp., USA SOURCE: Eur. Pat. Appl., 43 pp.

CODEN: EPXXDW

DOCUMENT TYPE: Patent LANGUAGE: English

FAMILY ACC. NUM. COUNT: 2

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
EP 531143 EP 531143	A2 A3	19930310 19931020	EP 1992-308022	19920904
R: DE, FR, US 5239405	Ä	19930824		19910906
US 5500760 JP 07216349 US 5611966	A A2 A	19960319 19950815 19970318	JP 1992-238612	19920827 19920907 19950601
US 5985184 US 6143209	A A	19991116 20001107	US 1997-956198	19971022 19990604
PRIORITY APPLN. INFO	).:		US 1992-935784 A	19910906 19920827
			US 1993-61742 A3	19920904 19930117 19950601
			US 1997-819652 B1	19970317 19971022

OTHER SOURCE(S): MARPAT 120:229898

AB Electrochemichromic solns. are described which comprise .gtoreq.1 anodic compd. which has had its valence state changed by contact with a redox agent, .gtoreq.1 cathodic compd. and a solvent; the redox potential of the anodic compd. is greater than that of the cathodic compd. when in contact with the solvent. Devices (e.g., adjustable mirrors) employing the solns. in conjunction with a cell provided with electrodes are also described.

IT 29260-45-5 30719-07-4

AUTHOR(S):

RL: PRP (Properties)

(electrochemichromic solns. contq.)

L25 ANSWER 8 OF 20 HCAPLUS COPYRIGHT 2003 ACS on STN

ACCESSION NUMBER: 1990:15675 HCAPLUS

DOCUMENT NUMBER: 112:15675

TITLE: Polarographic determination of lead and tin in lead

tin telluride (Pb1-xSnxTe) semiconductor alloys Kopanskaya, L. S.; Kiriyak, L. G.; Smelov, V. A.; Russu, V. G.; Vataman, I. I.; Slobodenyuk, K. I.

CORPORATE SOURCE: Inst. Khim., Kishinev, USSR

SOURCE: Zavodskaya Laboratoriya (1989), 55(4), 10-13

CODEN: ZVDLAU; ISSN: 0321-4265

DOCUMENT TYPE: Journal

LANGUAGE: Russian

Simultaneous detn. of Sn and Pb in Pb1-xSnxTe is based on formation of AB assocs. of Sn complexes of H2C2O4 and protonated leuco base of Azure I on the surface of a Hg drop. At the optimal conditions for polarog. detn. which included 0.5M H2C2O4 and 10-4M Azure I, potential peaks of Pb and Sn were EPb = -0.425 and ESn = -0.545 V (vs. SCE). Te did not interfere, and it can be detd. in the system from the peak ETe = -0.805 V. Also, Pb and Sn were isolated from the sample and detd. sep., which allowed confirmation of the accuracy of this method. Pb was selectively extd. as diethyldithiocarbamate at pH 10-11 (to assure full sepn. of Te(IV) and Sn(IV) with CHCl3). This was followed by polarog. of the ext. in the presence of 1M LiClO4 in EtOH contg. 1M HCl. The latter is necessary to decomp. the In(III) dopant which if present would ext. along with Pb. In these conditions the Pb peak is at EPb = -0.42 V (with resp. to Hg electrode). Sn was sepd. from the sample by copptn. on Be hydroxide in the presence of EDTA, followed by detn. using H2C2O4 and Azure I. 29260-45-5D, ion assoc. with tin oxalato complexes ΙT RL: PRP (Properties) (elec. potential of) L25 ANSWER 9 OF 20 HCAPLUS COPYRIGHT 2003 ACS on STN 1989:486027 HCAPLUS ACCESSION NUMBER: DOCUMENT NUMBER: 111:86027

Electrochemical properties of macromolecular bound TITLE:

ruthenium(II) complexes coated electrodes

Ramaraj, R.; Natarajan, P. AUTHOR(S):

Dep. Inorg. Chem., Univ. Madras, Madras, 600 025, CORPORATE SOURCE:

India

Indian Journal of Chemistry, Section A: Inorganic, SOURCE:

Physical, Theoretical & Analytical (1989), 28A(3),

187-96

CODEN: IJCADU; ISSN: 0376-4710

DOCUMENT TYPE: Journal LANGUAGE: English

Cyclic voltammetric behavior of electrodes coated with macromol. bound Ru complexes was investigated. Bis(bipyridine)dichlororuthenium(II) was condensed with poly(4-vinylpyridine) and its copolymers with methylolacrylamide. Macromol. Ru complexes contq. covalently bound thionine dye were prepd., and their electrochem. behavior was investigated. The no. of Ru complex centers bound to the macromol. chain was varied, and the redox behavior of the metal complex in the presence of

30719-07-4D, N-Methylthionine, derivs., reaction products with IΤ acrylamide-methylolacrylamide-vinylpyridine polymer, ruthenium bipyridine complexes

RL: PRP (Properties)

(platinum electrode coated with, cyclic voltammetry of)

L25 ANSWER 10 OF 20 HCAPLUS COPYRIGHT 2003 ACS on STN

ACCESSION NUMBER: 1988:551298 HCAPLUS

the macromol. chain was studied.

DOCUMENT NUMBER: 109:151298

Donnan approach to equilibrium sorption: interactions TITLE:

of cationic dyes with acrylic fibers

Alberghina, Gaetano; Chen, Shuilin; Fisichella, AUTHOR(S):

Salvatore; Iijima, Toshiro; McGregor, Ralph; Rohner,

Rolf M.; Zollinger, Heinrich

Univ. Catania, Catania, Italy CORPORATE SOURCE:

Textile Research Journal (1988), 58(6), 345-54 SOURCE:

CODEN: TRJOA9; ISSN: 0040-5175

DOCUMENT TYPE: Journal English LANGUAGE:

The sorption isotherms of 10 cationic dyes on a series of regular and porous acrylic fibers (e.g., Dralon X-100, Dunova, ATF 1017, RIH 1920, RIH 71600, and Dralon U-100) were detd. in detail and interpreted by means of

the Donnan theory, taking into account that the fibers used contain 2 types of ionizable groups, namely strongly acidic SO3H and OSO3H groups and weakly acidic CO2H groups. The relative magnitudes of the ionic distribution coeffs. for the dyes defined an "affinity series" which was essentially the same for all the fibers in spite of the known variations in fiber structure and porosity. This suggests that dye adsorption on internal surfaces or interfaces is not a major factor, and that the dye sorption is detd. primarily by the ion-exchange and acid-base characteristics of the polymer matrix, modified to some extent perhaps by the intrinsic water sorptions of the different polymer phases,. The distribution coeffs. correlate in an interesting manner with the mol. wts. and chem. structures of the dyes.

29260-45-5 ΙT

> RL: PEP (Physical, engineering or chemical process); PROC (Process) (sorption of, by acrylic fibers, Donnan equil. in relation to)

ANSWER 11 OF 20 HCAPLUS COPYRIGHT 2003 ACS on STN

1988:23242 HCAPLUS ACCESSION NUMBER:

DOCUMENT NUMBER: 108:23242

TITLE: Donnan approach to the equilibrium sorption

interactions of cationic dyes with acrylic fibers Alberghina, Gaetano; Chen, Shuilin; Fisichella, Salvatore; Iijima, Toshiro; McGregor, Ralph; Rohner, AUTHOR(S):

Rolf M.; Zollinger, Heinrich

Univ. Catania, Catania, Italy CORPORATE SOURCE:

Book of Papers - International Conference & SOURCE:

Exhibition, AATCC (1987) 216-19 CODEN: BPIAEQ; ISSN: 0892-2713

DOCUMENT TYPE: Journal English LANGUAGE:

The equil. sorption of 10 cationic dyes by porous and regular acrylic fibers was interpreted by a simple Donnan approach, based on the assumption that there were 2 different types of acidic groups in the fibers. The ionic distribution coeffs. (KD) for the dyes and the fibers were calcd. These coeffs. provided an indirect measure of the affinity of the dyes for the fibers. The values of KD were discussed in relation to the structure of the dyes and fibers and in relation to the effects of

salts, pH, and temp. 29260-45-5 IT

> RL: PEP (Physical, engineering or chemical process); PROC (Process) (sorption of, by acrylic fibers, detn. of, by Donnan approach)

L25 ANSWER 12 OF 20 HCAPLUS COPYRIGHT 2003 ACS on STN

ACCESSION NUMBER: 1984:475916 HCAPLUS

DOCUMENT NUMBER: 101:75916

TITLE: Electron-transfer coupling in microbial fuel cells.

1. Comparison of redox-mediator reduction rates and

respiratory rates of bacteria

Roller, Sibel D.; Bennetto, H. Peter; Delaney, Gerard AUTHOR(S):

M.; Mason, Jeremy R.; Stirling, John L.; Thurston,

Christopher F.

Biotech. Group, Queen Elizabeth Coll., London, W8 7AH, CORPORATE SOURCE:

Journal of Chemical Technology and Biotechnology, SOURCE:

> Biotechnology (1984), 34(1), 3-12 CODEN: JTBBD7; ISSN: 0264-3421

DOCUMENT TYPE: Journal LANGUAGE: English

Redox mediators promote electron transfer in microbial fuel cells. redn. of a range of redox mediators by bacteria was studied to identify effective mediator-organism combinations. Rates of redn. of mediator dyes by bacteria were measured spectrophotometrically at 30.degree. under anaerobic conditions for standardized concns. of organism, substrate, and

dye. The kinetics of dye redn. showed 2 general patterns: a simple, exponential curve or a complex curve with an initial linear rate followed by a faster exponential rate of redn. Dye-redn. rates were greater than rates of O consumption for several combinations of organism and redox dye. The use of these dyes as electron-transfer mediators in microbial fuel cells is discussed.

80297-51-4 IT

RL: USES (Uses)

(redox mediator, in microbial fuel cells, properties of)

L25 ANSWER 13 OF 20 HCAPLUS COPYRIGHT 2003 ACS on STN

ACCESSION NUMBER:

1984:475915 HCAPLUS

DOCUMENT NUMBER:

101:75915

TITLE:

Electron-transfer coupling in microbial fuel cells. Performance of fuel cells containing selected

microorganism-mediator-substrate combinations

AUTHOR(S):

Delaney, Gerard M.; Bennetto, H. Peter; Mason, Jeremy

R.; Roller, Sibel D.; Stirling, John L.; Thurston,

Christopher F.

CORPORATE SOURCE:

Biotech. Group, Queen Elizabeth Coll., London, W8 7AH,

UK

SOURCE:

Journal of Chemical Technology and Biotechnology,

Biotechnology (1984), 34(1), 13-27 CODEN: JTBBD7; ISSN: 0264-3421

DOCUMENT TYPE:

Journal

LANGUAGE:

English

Various phenoxazine, phenothiazine, phenazine, indophenol, and bipyridilium derivs. were tested for their effectiveness as redox mediators in microbial fuel cells contg. Alcaligenes eutrophus, Bacillus. subtilis, Escherichia coli, or Proteus vulgaris as the active biol. agent, and glucose [50-99-7] or succinate as the oxidizable substrate. A ferricyanide-Pt cathode was used. The open-circuit cell emf.'s increased with increasing neg. redox potential at pH 7 of the redox compds. Several of the redox agents worked well as mediators, maintaining steady currents over several hours, and thionine [581-64-6] was esp. effective in maintaining relatively high cell voltages when current was drawn from the cell. P. vulgaris, With thionine as mediator and glucose as substrate, showed the best performance in a fuel cell. This system was examd. in some detail under various conditions of external load to establish the effects of organism concn., mediator concn., and substrate addn.

80297-51-4 TT

RL: USES (Uses)

(redox mediator, performance of microbial fuel cells contg.)

L25 ANSWER 14 OF 20 HCAPLUS COPYRIGHT 2003 ACS on STN

ACCESSION NUMBER:

1982:173548 HCAPLUS

DOCUMENT NUMBER:

96:173548

TITLE:

Formation and extraction of tungstophosphate-thiazine dye compounds. Spectrophotometric determination of

phosphate

AUTHOR(S):

Sanchez-Pedreno, C.; Hernandez Cordoba, M.; Ortuno, J.

A.; Tudela, G. M.

CORPORATE SOURCE:

Spain

SOURCE:

Anales de la Universidad de Murcia, Ciencias (1981),

37(1-4), 91-106

CODEN: AUMCB5; ISSN: 0463-9847

DOCUMENT TYPE:

Journal

LANGUAGE:

Spanish

Although phosphate anions do not form extractable compds. with basic dyes, tungstophosphate assocs. with several dyes, and the assocs. can be extd. with org. solvents. The thiazine dye Azure B was used to  $\det$  0.2-10 ppm orthophosphate by forming the ion-assoc. with tungstophosphate, extg. into C6H6-cyclohexanone (1:1 by vol.), and measuring the absorbance at 645 nm.

The effects of all reagent concns., of H2SO4 concn., and of possible interfering species were examd. The calibration plot is linear for PO43detn. in 0.2N H2SO4, but curved in 0.02N H2SO4.

IT 29260-45-5D, tungstophosphate ion assoc.

> RL: PRP (Properties) (spectrum of)

ANSWER 15 OF 20 HCAPLUS COPYRIGHT 2003 ACS on STN

ACCESSION NUMBER: 1982:8116 HCAPLUS

DOCUMENT NUMBER: 96:8116

TITLE: New thiazine dyes for photogalvanic cells

Albery, W. John; Bartlett, Philip N.; Davies, John P.; AUTHOR(S):

Foulds, Andrew W.; Hillman, A. Robert; Bachiller,

Fernando Souto

CORPORATE SOURCE: Dep. Chem., Imp. Coll. Sci. Technol., London, SW7 2AY,

UK

Faraday Discussions of the Chemical Society (1981), SOURCE:

70, 341-57

CODEN: FDCSB7; ISSN: 0301-7249

DOCUMENT TYPE: Journal LANGUAGE: English

Six new modified thiazine dyes were prepd. and studied in relation to the photochem. and kinetic requirements of a photogalvanic cell for solar energy conversion, since the efficiency of the Fe thionine photogalvanic cell for solar energy conversion is severely limited by the insoly. of thionine. A linear free energy relation exists between the rate const. for the thermal back reaction and the std. electrode potential of the thiazine dye. The implications of this relation are explored, together with the effect of pH, Fe(III) concn., and soly. on the efficiency of the cell. For efficient energy conversion it is necessary to have a selective electrode; the thionine-coated electrode is suitable for the cell.

TΤ 80297-51-4D, sulfonated

RL: USES (Uses)

(electrolytes contg. iron and, for photogalvanic cells)

ANSWER 16 OF 20 HCAPLUS COPYRIGHT 2003 ACS on STN

1980:597095 HCAPLUS ACCESSION NUMBER:

DOCUMENT NUMBER: 93:197095

TITLE: Conditions for quantitative formation of

molybdosilicic acid-thiazine dye compounds

Mirzoyan, F. B.; Tarayan, V. M. AUTHOR(S):

CORPORATE SOURCE: Inst. Gen. Inorg. Chem., Yerevan, USSR

Zhurnal Analiticheskoi Khimii (1980), 35(7), 1293-300 SOURCE:

CODEN: ZAKHA8; ISSN: 0044-4502

DOCUMENT TYPE: Journal

LANGUAGE: Russian

The reaction between molybdosilicic acid (I) and the thiazine dyes, methylene blue, methylene green, and Azur I, was studied. The optimum acidity range for formation of I is pH 1.5-4.1. This range does not depend on the nature of the basic dye, but widens with the initial molybdate concn. After formation of I, the acidity of the soln. was increased to suppress completely the formation of basic dye isopolymolybdates and to sep. quant. the dye-I compds. The pH range is wider in the presence of oxalates, which mask the molybdate and allow the sepn. of the dye-I compds. under acidity conditions optimum for I formation. A highly sensitivity photometric method was developed for the detn. of Si as a I-methylene blue compd. (molar absorptivity = 4.3 .times. 105).

29260-45-5DP, compd. with molybdosilicic acid IT

RL: FORM (Formation, nonpreparative); PREP (Preparation)

(formation of, pH effect on)

L25 ANSWER 17 OF 20 HCAPLUS COPYRIGHT 2003 ACS on STN

ACCESSION NUMBER: 1979:585999 HCAPLUS

DOCUMENT NUMBER: 91:185999

TITLE: Reaction of molybdogermanic acid with thiazine dyes AUTHOR(S): Mirzoyan, F. V.; Tarayan, V. M.; Airiyan, E. Kh.

CORPORATE SOURCE: Inst. Obshch. Neorg. Khim., Yerevan, USSR

SOURCE: Armyanskii Khimicheskii Zhurnal (1979), 32(2), 106-12

CODEN: AYKZAN; ISSN: 0515-9628

DOCUMENT TYPE: Journal LANGUAGE: Russian

AB Ge was detd. by spectrophotometry by measuring the absorbance of the 1:4 complexes of molybdogermanic acid with dimethylthionine (I) or methylene green (II) in Me2CO at 610 or 650 nm, resp. (molar absorptivities 2.45 .times. 105 and 2.30 .times. 105, resp.). Beer's law was obeyed for 2 .times. 10-7-1.4 .times. 10-5 and 2 .times. 10-7-2.4 .times. 10-5M solns. contg. I and II, resp. The error was .ltoreq.3%.the sample soln. (2 mL) contg. 0.14-10.2 .mu.g Ge4+ was treated with HNO3 to adjust pH to 1.5-4.3 and it was reacted with 0.5 mL 0.024M Na2MoO4 for 10-15 min. Then 0.5 mL 0.2M Na2C2O4 and 0.5 mL 0.1% I or II were added and the vol. was adjusted to 10 mL. The ppt. was sepd. by centrifuging and it was dissolved in Me2CO.

IT 39093-22-6D, molybdogermanic complex

RL: PRP (Properties) (spectra of)

L25 ANSWER 18 OF 20 HCAPLUS COPYRIGHT 2003 ACS on STN

ACCESSION NUMBER: 1978:163374 HCAPLUS

DOCUMENT NUMBER: 88:163374

TITLE: Spectrophotometric determination of boron in thorium

sulfate.

AUTHOR(S): Federgrun, L.; Abrao, A.

CORPORATE SOURCE: Div. Eng. Quim., Inst. Energ. At., Sao Paulo, Brazil

SOURCE: Report (1976), IEA-420, 6 pp. Avail.: INIS

From: INIS Atomindex 1978, 9(3), Abstr. No. 354290

DOCUMENT TYPE: Report LANGUAGE: Portuguese

AB The detn. of microquantities of B in nuclear grade Th sulfate is based on the extn. of the BF4- ion assocd. with monomethylthionine (MMT) into 1,2-dichloroethane. The quant. sepn. of Th is mandatory to avoid the pptn. of ThF4. The Th sulfate is dissolved by using a strong cation exchanger, Th4+ being totally retained on the resin. B is then detd. in the effluent. The procedure allows the detn. of 0.2-10.0 .mu.g B with a max. error of 10%. Th sulfate samples contg. 0.2-2.0 .mu.g B/g Th were analyzed.

IT 30719-07-4

RL: ANST (Analytical study)

(in detn. of boron by extn. and spectrophotometry)

L25 ANSWER 19 OF 20 HCAPLUS COPYRIGHT 2003 ACS on STN

ACCESSION NUMBER: 1973:9548 HCAPLUS

DOCUMENT NUMBER: 78:9548

TITLE: Theoretical study of the spectra and protonation site

of thiazine dyes. II. Triplet state

AUTHOR(S): Rayez, J. C.; Chalvet, O.

CORPORATE SOURCE: Lab. Chim. Phys. A, Univ. Bordeaux, Talence, Fr. SOURCE: Journal de Chimie Physique et de Physico-Chimie

Biologique (1972), 69(10), 1537-43

CODEN: JCPBAN; ISSN: 0021-7689

DOCUMENT TYPE: Journal LANGUAGE: French

AB A configuration interaction calcn. within the Pariser-Parr-Pople approxn. indicates that the lowest triplet and lowest singlet states of thionine have the same configuration rather than different configurations as

suggested by U. Sommer and H. E. A. Kramer (1971). The 1st triplet state

of basic and acid thiazine dyes are involved in the same electronic transition (9 .fwdarw. 10) as are the 1st singlet states. The protonation site of these basic dyes is the intracyclic N atom in these states.

IT 29260-45-5 30719-07-4 39093-22-6

RL: PRP (Properties)

(protonation and spectrum of, configuration interaction in relation to)

L25 ANSWER 20 OF 20 HCAPLUS COPYRIGHT 2003 ACS on STN

ACCESSION NUMBER:

1971:44920 HCAPLUS

DOCUMENT NUMBER:

74:44920

TITLE:

Quantitative determination of boron in glasses used as

encapsulants for electronic devices

AUTHOR(S):

Govdish, B. L.

CORPORATE SOURCE:

RCA Lab., Princeton, NJ, USA

SOURCE:

Microchemical Journal (1970), 15(4), 572-8

CODEN: MICJAN; ISSN: 0026-265X

DOCUMENT TYPE:

Journal English

LANGUAGE:

N-Methylthionine (Azure C) reacts with BF4- to form a colored complex in a 0.5N H2SO4 medium. This colored complex is extractable with dichloroethane or a mixt. of dichloroethane-dichloropropane, with a max. absorbance at 660 m.mu. Since it is necessary to dissolve the borosilicate glass with HF, the presence of fluoride creates a problem when attempting to utilize other spectrophotometric methods for B, e.g., hydroxy-anthraquinone and anthraquinonylamine reagents. In the proposed method, fluoride is necessary because Azure C reacts only with the BF4-complex; this makes the method very selective for the detn. of B in the presence of fluoride without prior sepn. This method has been successfully applied to the detn. of B in glasses used as encapsulants.

IT 30719-07-4

RL: USES (Uses)

(in boron detn., in glass)

=>

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=> fil caold

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FILE LAST UPDATED: 01 May 1997 (19970501/UP)

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This file supports REGISTRY for direct browsing and searching of all substance data from the REGISTRY file. Enter HELP FIRST for more information.

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L26

0 L24

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DICTIONARY FILE UPDATES: 22 SEP 2003 HIGHEST RN 591204-55-6

TSCA INFORMATION NOW CURRENT THROUGH JULY 14, 2003

Please note that search-term pricing does apply when conducting SmartSELECT searches.

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Experimental and calculated property data are now available. See HELP PROPERTIES for more information. See STNote 27, Searching Properties in the CAS Registry File, for complete details: http://www.cas.org/ONLINE/STN/STNOTES/stnotes27.pdf

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L24 ANSWER 1 OF 10 REGISTRY COPYRIGHT 2003 ACS on STN

RN 352005-63-1 REGISTRY

CN Phenothiazin-5-ium, 3-[(4-amino-3-methylphenyl)amino]-2-methyl-7-(methylamino)- (9CI) (CA INDEX NAME)

FS 3D CONCORD

MF C21 H21 N4 S

SR CA

LC STN Files: CA, CAPLUS

1 REFERENCES IN FILE CA (1907 TO DATE)
1 REFERENCES IN FILE CAPLUS (1907 TO DATE)

REFERENCE 1: 135:134287

L24 ANSWER 2 OF 10 REGISTRY COPYRIGHT 2003 ACS on STN

RN 352005-61-9 REGISTRY

CN Phenothiazin-5-ium, 7-amino-2-methyl-3-(methylamino)- (9CI) (CA INDEX NAME)

FS 3D CONCORD

MF C14 H14 N3 S

SR CA

LC STN Files: CA, CAPLUS

1 REFERENCES IN FILE CA (1907 TO DATE)

1 REFERENCES IN FILE CAPLUS (1907 TO DATE)

REFERENCE 1: 135:134287

L24 ANSWER 3 OF 10 REGISTRY COPYRIGHT 2003 ACS on STN

RN 352005-60-8 REGISTRY

CN Phenothiazin-5-ium, 2-methyl-3,7-bis(methylamino)- (9CI) (CA INDEX NAME)

FS 3D CONCORD

MF C15 H16 N3 S

SR CA

LC STN Files: CA, CAPLUS

1 REFERENCES IN FILE CA (1907 TO DATE)

1 REFERENCES IN FILE CAPLUS (1907 TO DATE)

REFERENCE 1: 135:134287

L24 ANSWER 4 OF 10 REGISTRY COPYRIGHT 2003 ACS on STN

RN 352005-59-5 REGISTRY

CN Phenothiazin-5-ium, 3-amino- (9CI) (CA INDEX NAME)

FS 3D CONCORD

MF C12 H9 N2 S

SR CA

LC STN Files: CA, CAPLUS

1 REFERENCES IN FILE CA (1907 TO DATE)

1 REFERENCES TO NON-SPECIFIC DERIVATIVES IN FILE CA

1 REFERENCES IN FILE CAPLUS (1907 TO DATE)

REFERENCE 1: 135:134287

L24 ANSWER 5 OF 10 REGISTRY COPYRIGHT 2003 ACS on STN

RN 153813-86-6 REGISTRY

CN Phenothiazin-5-ium, 3-(dimethylamino)- (9CI) (CA INDEX NAME)

FS 3D CONCORD

MF C14 H13 N2 S

CI COM SR CA

L24 ANSWER 6 OF 10 REGISTRY COPYRIGHT 2003 ACS on STN

RN 80297-51-4 REGISTRY

CN Phenothiazin-5-ium, 3,7-bis(methylamino)disulfo-, inner salt, ion(1-)

(9CI) (CA INDEX NAME)

MF C14 H12 N3 O6 S3

CI IDS

LC STN Files: CA, CAPLUS

3 REFERENCES IN FILE CA (1907 TO DATE)

1 REFERENCES TO NON-SPECIFIC DERIVATIVES IN FILE CA

3 REFERENCES IN FILE CAPLUS (1907 TO DATE)

REFERENCE 1: 101:75916

REFERENCE 2: 101:75915

REFERENCE 3: 96:8116

L24 ANSWER 7 OF 10 REGISTRY COPYRIGHT 2003 ACS on STN

RN 47078-64-8 REGISTRY

CN Phenothiazin-5-ium, 7-(dimethylamino)-2-methyl-3-(methylamino)- (9CI) (CA

INDEX NAME)

FS 3D CONCORD

MF C16 H18 N3 S

CI COM

LC STN Files: CA, CAPLUS

1 REFERENCES IN FILE CA (1907 TO DATE)

1 REFERENCES IN FILE CAPLUS (1907 TO DATE)

1: 135:134287 REFERENCE

ANSWER 8 OF 10 REGISTRY COPYRIGHT 2003 ACS on STN L24

RN 39093-22-6 REGISTRY

Phenothiazin-5-ium, 3,7-bis(methylamino)- (9CI) (CA INDEX NAME) CN

FS. 3D CONCORD

MF C14 H14 N3 S

CI COM

LC STN Files: BEILSTEIN\*, CA, CAPLUS

(\*File contains numerically searchable property data)

3 REFERENCES IN FILE CA (1907 TO DATE)

2 REFERENCES TO NON-SPECIFIC DERIVATIVES IN FILE CA

3 REFERENCES IN FILE CAPLUS (1907 TO DATE)

1: 131:334255 REFERENCE

2: 91:185999 REFERENCE

REFERENCE 3: 78:9548

L24 ANSWER 9 OF 10 REGISTRY COPYRIGHT 2003 ACS on STN

30719-07-4 REGISTRY RN

Phenothiazin-5-ium, 3-amino-7-(methylamino)- (8CI, 9CI) (CA INDEX NAME) CN

OTHER NAMES:

Monomethylthionine CN

CN N-Methylthionine

3D CONCORD FS

C13 H12 N3 S MF

CI COM

LC STN Files: BEILSTEIN\*, CA, CAPLUS, CSCHEM, USPATFULL (\*File contains numerically searchable property data)

$$H_2N$$

8 REFERENCES IN FILE CA (1907 TO DATE)

2 REFERENCES TO NON-SPECIFIC DERIVATIVES IN FILE CA

8 REFERENCES IN FILE CAPLUS (1907 TO DATE)

REFERENCE 1: 131:334255

2: 125:341110 REFERENCE

3: 124:328070 REFERENCE

4: 120:229898 REFERENCE

5: 111:86027 REFERENCE

REFERENCE 6: 88:163374

REFERENCE 7: 78:9548

REFERENCE 8: 74:44920

L24 ANSWER 10 OF 10 REGISTRY COPYRIGHT 2003 ACS on STN

RN 29260-45-5 REGISTRY

CN Phenothiazin-5-ium, 3-(dimethylamino)-7-(methylamino)- (9CI) (CA INDEX

NAME)

FS 3D CONCORD

DR 56109-47-8, 155614-09-8

MF C15 H16 N3 S

CI COM

LC STN Files: BEILSTEIN\*, CA, CAPLUS, TOXCENTER, USPATFULL (\*File contains numerically searchable property data)

12 REFERENCES IN FILE CA (1907 TO DATE)

4 REFERENCES TO NON-SPECIFIC DERIVATIVES IN FILE CA

12 REFERENCES IN FILE CAPLUS (1907 TO DATE)

REFERENCE 1: 136:226765

REFERENCE 2: 131:334255

REFERENCE 3: 125:341110

REFERENCE 4: 125:329347

REFERENCE 5: 124:328070

REFERENCE 6: 120:229898

REFERENCE 7: 112:15675

REFERENCE 8: 109:151298

REFERENCE 9: 108:23242

REFERENCE 10: 96:173548